



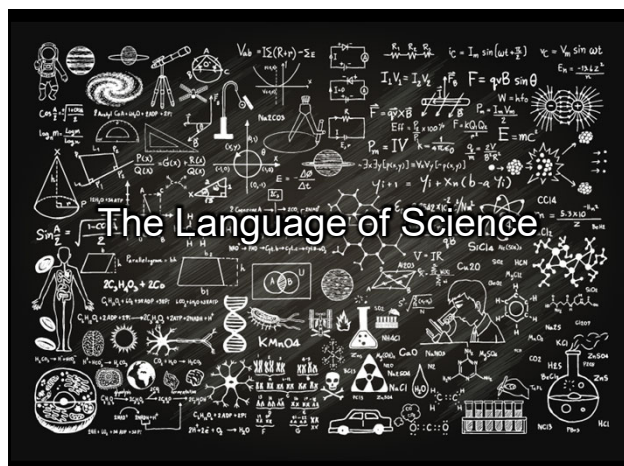
What is science?

- Science is a process by which we try to understand how the natural world works and how it came to be that way.



- Scientists investigate and observe phenomena in the natural world to gather evidence.
- This evidence is analyzed to try to make sense of what is happening.
- The scientist then provides an explanation or theory as to why or how the phenomena occurs.
- This explanation is always considered tentative as there are limitations of what we know and what we can observe.

- All explanations are not equally valid. Explanations that best fit the data are more valid than others.
- It must be possible to disprove an explanation.
- Scientific explanations cannot be based merely upon personal opinion, popular vote, belief, or judgment.
- Science is a human endeavor. Therefore, bias, opinion, ego, and other human factors will come into play. The task is to try to remove those factors from a scientific explanation.



Units

- The measurements of physical quantities are expressed in terms of units.

Physical Quantity	Units
time	second (s)
mass	kilogram (kg)
distance	meter (m)
volume	liter (L)
speed	meters/second (m/s)
temperature	Celsius (°C)

SI Prefixes

Small			Large		
centi	c	10^{-2}	kilo	k	10^3
milli	m	10^{-3}	mega	M	10^6
micro	μ	10^{-6}	giga	G	10^9
nano	n	10^{-9}	terra	T	10^{12}
pico	p	10^{-12}			

Converting Units

- Calculations are done using base units.
- To convert to base units, multiply the value by the appropriate multiplier.

$$2 \text{ nm} = ? \text{ m}$$

The multiplier for **nano** is 10^{-9} .

$$2 \text{ nm} = 2 \times 10^{-9} \text{ m}$$

- To convert from base units to a prefixed value, divide by the appropriate multiplier.

$$5000 \text{ m} = ? \text{ km}$$

The multiplier for **kilo** is 10^3 .

$$\frac{5000 \text{ m}}{10^3} = 5 \text{ km}$$



Examples

$$50 \mu\text{m} = \underline{50 \times 10^{-6}} \text{ m}$$

$$250 \text{ g} = \underline{0.25} \text{ kg}$$

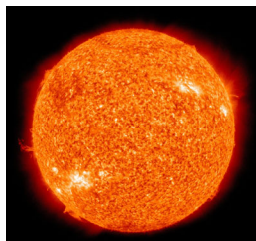
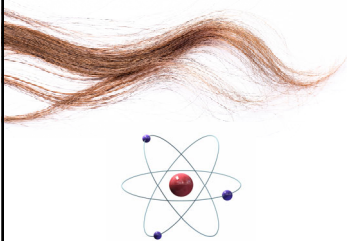
$$\frac{250}{10^3}$$

Scientific Notion

- Many measurements we encounter are values that are easily understood and manipulated.
 - Volume of a soda can = 355 mL
 - Distance from Winnipeg to Toronto = 2000 km



- But there also are extreme values.
 - Width of a human hair = 0.00005 m
 - Electron radius = 0.000000000000047 m
 - Mass of the sun = 20000000000000000000000000000000 kg



•A shorthand method of writing very small and very large numbers is called scientific notation, in which we express numbers in terms of exponents of 10.

•Scientific notation follows the general format $a \times 10^n$. Where a is a decimal number and n is an integer.

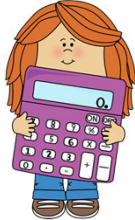
• 1.67×10^{-27}

• 5.97×10^{24}

•To write a number in scientific notation, move the decimal point to the right of the first digit in the number.

•Count the number of places that you moved the decimal point.

•The number of places moves is the exponent.



•For large numbers, the decimal moves to the left and the exponent will be positive.

123.000.000.000.

$$1.23 \times 10^{11}$$

• For small numbers, the decimal moves to the right and exponent will be negative.

0.000.000.001.23

$$1.23 \times 10^{-9}$$

Examples

$$250\,000\,000\text{ m} = \underline{2.5 \times 10^8 \text{ m}}$$

$$0.000\,006\,8\text{ kg} = \underline{6.8 \times 10^{-6} \text{ kg}}$$

Uncertainty

- There is an inherent uncertainty in any measurement. It is caused by random fluctuations within a system.
- This uncertainty cannot be eliminated but it can be measured and reduced by conducting repeated observations of a specific event.

- The amount of uncertainty in a measurement is expressed by +/- notation.

1.23 ± 0.05

measurement value \nearrow \nwarrow uncertainty

The value of the measurement is between 1.18 ($1.23 - 0.05$) and 1.28 ($1.23 + 0.05$).
